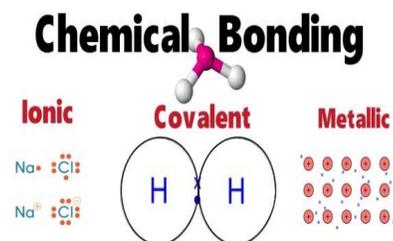
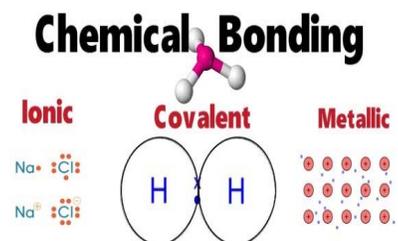


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- Refer the modern periodic table and answer the following questions.
  - The elements placed in group number 18 are called.....
  - Alkali and alkaline earth metals are collectively called .....block metals.
  - The general configuration for halogens is.....
  - Name a p-block element which is a gas other than a noble gas or a halogen.
  - Name the groups that comprise the 's' block of elements.
  - Element number 118 has not yet been established, to which block, will it belong?
  - How many elements should be there in total if all the 7s, 7p, 6d and 5f, blocks are to be full?
- Describe the variation of electron gain enthalpy and ionization enthalpy in the periodic table.
- Which one have lower first ionization enthalpy and why?
  - Na or Ca
  - K or Ar
  - Na<sup>+</sup> or Na
- Which one have high electron gain enthalpy and why?
  - O<sup>-</sup> or O<sup>2-</sup>
  - O<sup>-</sup> or S
  - N<sup>-</sup> or P
- Define the following:
  - Electron gain enthalpy
  - Ionization enthalpy
  - Ionic radius
  - Electronegativity.
- What is electronegativity? How is it related to the type of bond formed?
- Why is the electron gain enthalpy of Cl more in negative value as compared to that of F?
- Define ionic radii. Arrange these ionic species in order of decreasing ionic radii.
  - Na<sup>+</sup>, Mg<sup>2+</sup>, K<sup>+</sup>, Al<sup>3+</sup>
  - N<sup>3-</sup>, O<sup>2-</sup>, F<sup>-</sup>, Br<sup>-</sup>
- Explain Why
  - First ionization enthalpy of N is more than O.
  - Second electron gain enthalpy of O is negative.
  - All transition elements are d-block elements but all d-block elements are not transition elements.
  - N has positive electron gain enthalpy but O has negative electron gain enthalpy.

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10. Pl consider this part of periodic table and answer below given questions

Group Period	1	2	d-block Elements	13	14	15	16	17	18
2		A			M	F	G	X	
3	B							Y	
4	R		S					Z	
5	O			D					
6									Q

- Which one may be a transition element?
- What is the Family name of elements, XYZ?
- Which one is most reactive Metal?
- Which one is most reactive non metal?
- Which one is also known as chalcogens?
- Which one is alkali metal?
- Which one is alkaline earth metal?
- What is the valency of Q, X, N & P?
- What is the formula of compound formed by reaction of A & F?
- Which one has smallest and which one has biggest size?

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